Innovative Method to Produce 99 Mo for Medical Diagnosis without Uranium - Development of ⁹⁹Mo Production Technique with Molybdate Solution -

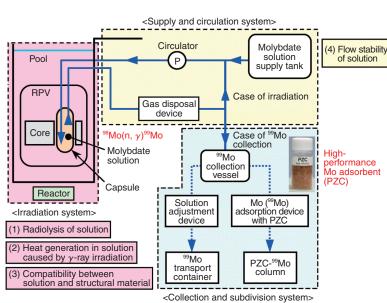


Fig.14-17 Outline of systems for 99Mo production by solution irradiation method, and evaluation items

In the irradiation system, a molybdate solution in a capsule installed in a reactor core is irradiated with neutrons, and 99Mo is generated. In the supply and circulation system, this solution is supplied to the capsule, and then circulated back. In the collection and subdivision system, the solution including generated 99Mo is collected from the capsule, and this solution is purified and put into transport vessels. (1) to (4) are evaluation items for two of these systems.

Technetium-99m (99mTc) is the most widely used radiopharmaceutical for medical diagnosis. The raw material of 99mTc is molybdenum-99 (99Mo), and all 99Mo used in Japan is imported from foreign countries. However, a problem has emerged that the supply of 99Mo is unstable due to aging production facilities. Therefore, the stable supply and production of 99Mo is needed in Japan.

⁹⁹Mo has been produced mainly by the fission method using highly enriched uranium. In Japan, 99Mo production by the fission method is difficult because of nuclear nonproliferation requirements and radioactive Therefore, taking a hint from the shipment form of ⁹⁹Mo, the solution irradiation method was invented (Fig.14-17). In this method, a solution of molybdate that has high solubility is irradiated with neutrons in a nuclear reactor. 99Mo is thus produced by the 98 Mo (n, γ) 99 Mo reaction and is collected with the high-performance Mo adsorbent PZC. This method has three advantages: 99Mo production rate can be increased simply by increasing the volume of the irradiation target, the processes required to produce a shipment are simpler, and the amount of radioactive waste is much smaller than that of the fission method.

As the molybdate solution used in the solution irradiation

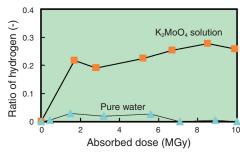


Fig.14-18 Relationship between absorbed dose of K₂MoO₄ solution and ratio of hydrogen in generated gas The ratio of hydrogen in the radiolysis gas generated from K₂MoO₄ solution (concentration: 58wt%, temperature: 80°C) was investigated. As a result, it was found that this solution generates more hydrogen than pure water does.

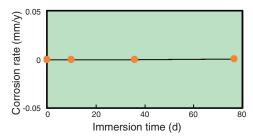


Fig.14-19 Relationship between immersion time and corrosion rate of SS304

SS304 was immersed in a K_2MoO_4 solution (the concentration: 58wt%, the temperature: 80°C), and the compatibility between SS304 and the solution was investigated. As a result, it was found that SS304 has no corrosion after 80 days of immersion.

method, potassium molybdate (K₂MoO₄) solution which has a high concentration of molybdenum was selected to increase the ⁹⁹Mo production rate. For the investigation of its suitability as the irradiation target, γ -ray irradiation tests of the solution were carried out. The test results indicated: (1) the ratio of hydrogen in the gas generated from the solution is 20 times higher than that from pure water (Fig.14-18); however, this is not a serious problem for the gas disposal device installed into the 99Mo production system, (2) the increase in calorific value of the solution upon γ -ray irradiation is about the same as that of pure water, (3) the compatibility (nonreactivity) between the solution and stainless steel material of the capsule is very good (Fig.14-19), (4) the solution flows stably in a pipe 4mm in inner diameter and 30m in length without precipitation.

From the above results, it is clear that the K₂MoO₄ solution is suitable as the irradiation target for the 99Mo production system by the solution irradiation method. In addition, it is evaluated that over 20% of the domestic demand for ⁹⁹Mo (88.8TBq/week) can be provided by this system using JMTR. Based on this obtained data, plans for the practical application of this system are to be considered.

Reference

Inaba, Y. et al., Development of 99Mo Production Technique by Solution Irradiation Method; Characterization of Aqueous Molybdate Solutions, Nippon Genshiryoku Gakkai Wabun Ronbunshi, vol.8, no.2, 2009, p.142-153 (in Japanese).