## 4-5 Metal Oxides Become Active Catalysts When Irradiated by Electron-Beams!

— Enhancement of the Oxidation of Organic Substances with Electron-Beam Irradiated  $\gamma$ -Alumina —

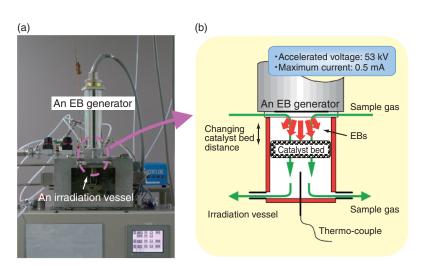


Fig.4-11 (a): An EB irradiation/catalyst system (external view), (b): an irradiation vessel for a gas stream (interior view)

This system consists of an EB generator, an irradiation vessel, catalysts, and gas analyzers. A sample gas containing organic substances was irradiated with EBs and then passed through a catalyst bed. The energy directly delivered to the catalyst bed was changed from zero to 50% of total EB energy by placing it at different distances from an EB irradiation window.

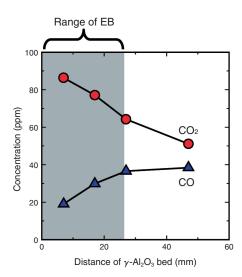


Fig.4-12 The concentrations of CO $_2$  and CO in 50-ppm xylene/air mixture irradiated at a dose of 10 kGy when  $\gamma$ -Al $_2$ O $_3$  bed was placed at various distances

10-kGy EB irradiation without a  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> bed produced 28.9-ppmv CO<sub>2</sub> and 38.5-ppmv CO. Placement of a  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> bed at a distance of 7mm caused higher production of CO<sub>2</sub> with less CO production.

Emission of various organic substances from painting factories causes photochemical smog. We have developed a gas purification system that can oxidize organic substances into  $CO_2$  and CO using electron beam (EB) irradiation. In this treatment, such organic substances should be preferentially oxidized into  $CO_2$  without producing toxic CO. We have recently focused on the development of EB use with catalysts for this purpose and found that a  $\gamma$ -alumina ( $\gamma$ -Al $_2O_3$ ) bed exhibited the best catalytic activity together with an electron accelerator, especially when its surface is directly irradiated with EBs.

The EB irradiation involved a variety of chemical reactions. For the understanding of complicated radiation-induced catalytic reactions, we manufactured an EB irradiation/catalyst system that can irradiate a sample gas in the presence of a catalyst bed placed at various distances from an EB-irradiation window. (Fig.4-11) The catalytic oxidation of o-xylene in air was studied using a  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> bed as a catalyst. The results were summarized as follows;

(1) Higher CO<sub>2</sub> concentration was obtained by an EB/ catalytic treatment.

- (2) Irradiation products of xylene oxidized exclusively into  $CO_2$  over the  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> surface.
- (3) The formation of  $CO_2$  was enhanced and the production of toxic CO was suppressed when a  $\gamma$ - $Al_2O_3$  bed was directly irradiated with EBs, namely placed in an EB-induced plasma. (Fig.4-12).

Our previous studies demonstrated that the placement of common catalysts such as  $TiO_2$  and  $MnO_2$  in an EB-irradiation space led to the decrease of  $CO_2$  concentration due to suppression of gas-phase oxidation reactions. On the other hand, the placement of  $\gamma$ -Al $_2O_3$  in the EB-irradiation space caused higher-concentration of  $CO_2$ . Such catalytic oxidation activity was found to be due to the exclusive oxidation of the irradiation products of xylene over EB-irradiating  $\gamma$ -Al $_2O_3$  surface into  $CO_2$ .

In general, metal oxides, such as  $\gamma$ -Al<sub>2</sub>O<sub>3</sub>, are made into chemical catalysts by loading noble metals on the surface. The present results are important not only for an EB-gas purification technology but also for production of a new catalyst which does not load any noble-metals.

## Reference

Hakoda, T. et al., Oxidation of Xylene and Its Irradiation Byproducts Using an Electron-Beam Irradiating a  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> Bed, Journal of Physics D: Applied Physics, vol.41, no.15, 2008, p.155202-1-155202-7.