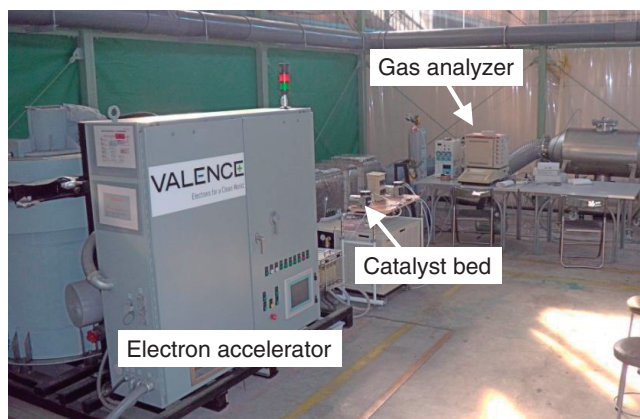


## 4-10 Purification of Factory Waste Gases Using By-Product Ozone — Electron Beam/Catalyst System for Purification of Waste Gases under Practical Gas-Flow Conditions —

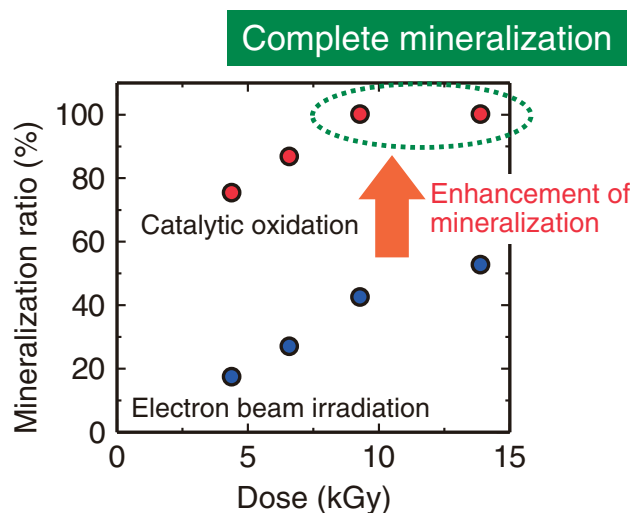


**Fig.4-19 Electron beam/ozonolysis catalyst (MnO<sub>2</sub>) system**

Toxic organics in an airstream (gas flow rate: 500 m<sup>3</sup>/h; gas velocity: 5 m/s) are oxidized into adhesive products under electron beam irradiation. The products are mineralized into CO<sub>2</sub> using active oxygen species produced from by-product ozone over an ozonolysis catalyst.

The presence of paint-solvent organics in waste gases emitted from painting factories has led to the production of photochemical oxidants in the atmosphere. We have developed a waste-gas purification system that can oxidize such organics into CO<sub>2</sub> by electron beam (EB) irradiation combined with catalytic oxidation. In previous studies, we found that an ozonolysis catalyst was the best matching catalyst since active oxygen species produced from the decomposition of by-product ozone over the catalyst under EB irradiation are effective in oxidizing organics. For the practical use of the purification system in factories emitting waste gases (flow rate:  $1 \times 10^4$  to  $2 \times 10^4$  m<sup>3</sup>/h; gas velocity: ~5 m/s), the effectiveness of the purification system should be evaluated under pilot-scale conditions (flow rate: 500~1000 m<sup>3</sup>/h; gas velocity: ~5 m/s).

In the present study, an EB/ozonolysis catalyst system was constructed for conducting a pilot-scale test for the mineralization of organics in an airstream (Fig.4-19). This system consists of a compact electron accelerator and a



**Fig.4-20 Enhancement of mineralization of toxic organics using electron beam/catalyst system**

A mixture of xylene and toluene (each with a concentration of 5 ppmv) in an airstream was mineralized into CO<sub>x</sub> (CO<sub>2</sub> and CO) under electron beam irradiation with and without catalytic oxidation. The mineralization ratio was obtained from the ratio of the carbon concentration of CO<sub>x</sub> to that of the input organics.

honeycomb-type manganese dioxide (MnO<sub>2</sub>) bed to efficiently decompose ozone in a turbulent airstream. The catalyst was heated to 100 °C for the desorption of the EB-generated nitric acid to deactivate the catalyst.

An airstream containing a mixture of xylene and toluene (concentration of each compound: 5 ppmv; flow rate: 500 m<sup>3</sup>/h; gas velocity: ~5 m/s) was introduced in the EB/MnO<sub>2</sub> system. The mineralization of these organics into CO<sub>2</sub> and CO was examined in the presence and absence of the catalyst. The presence of the catalyst enhanced the mineralization of unreacted xylene/toluene and their irradiation products: the mineralization ratio increased from 42% to 100% when the catalyst was used with an irradiation dose of 9.3 kGy (Fig.4-20).

In view of the present results, we intend applying this purification system to actual waste gases in cooperation with chemical engineering companies for protecting the earth's environment.

### Reference

Hakoda, T. et al., An Electron-Beam Irradiation/Catalytic Oxidation System for Purification of Aromatic Hydrocarbons/Air Mixture under Practical Gas-Flow Condition, *Industrial & Engineering Chemistry Research*, vol.49, no.12, 2010, p.5517-5522.