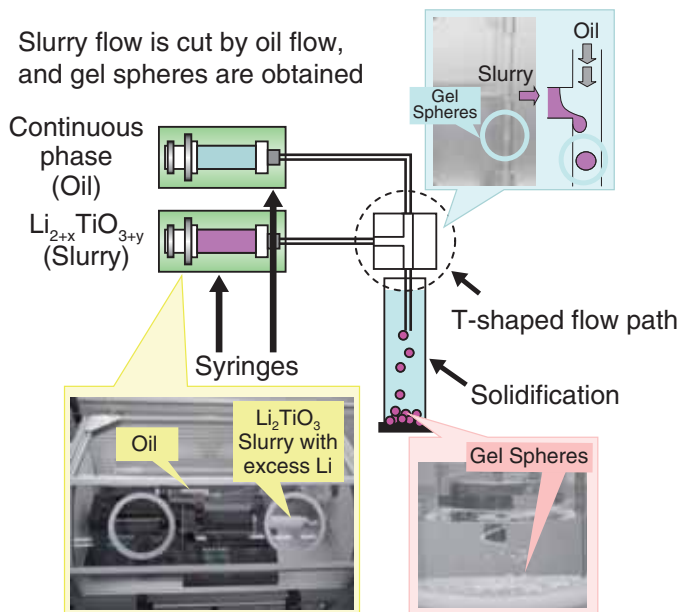


## 9-11 Stably Supplying Fuel Tritium to Fusion Reactors

- Establishment of a New Pebble Fabrication Technology for an Advanced Tritium Breeder -



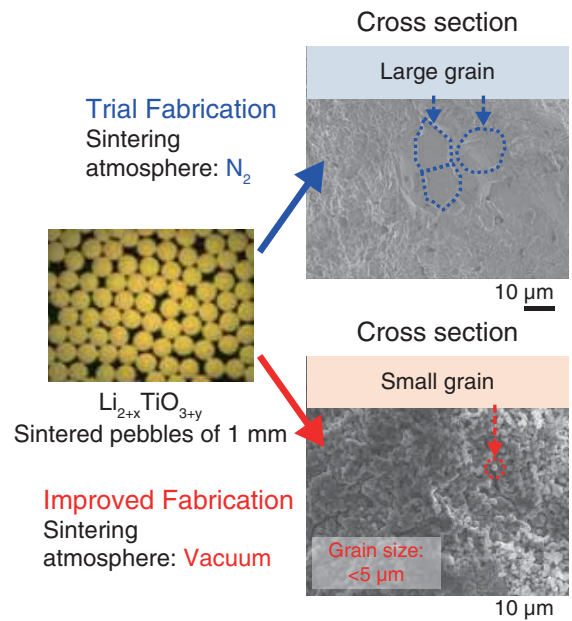
**Fig.9-27 Pebble fabrication of tritium breeders using an emulsion method**

The flow of the  $\text{Li}_2\text{TiO}_3$  slurry with excess Li ( $\text{Li}_{2+x}\text{TiO}_{3+y}$ ) is cut by an oil flow from an oil-filled syringe. The emulsion method is suitable for mass production of  $\text{Li}_{2+x}\text{TiO}_{3+y}$  pebbles.

Fusion reactors need deuterium (D) and tritium (T) as their fuel. Since tritium does not exist in nature, it is necessary to produce tritium in a reactor by neutron irradiation of lithium (Li). Lithium titanate ( $\text{Li}_2\text{TiO}_3$ ) is one of the most promising candidates among tritium breeders because of its tritium release characteristics. However, the mass of Li in the breeders decreases in a hydrogen atmosphere because of Li evaporation and Li burn-up.

To prevent the mass decrease of Li at high temperatures,  $\text{Li}_2\text{TiO}_3$  with excess Li ( $\text{Li}_{2+x}\text{TiO}_{3+y}$ ) has been developed as an advanced tritium breeder. The emulsion method as a pebble fabrication technique for  $\text{Li}_{2+x}\text{TiO}_{3+y}$  was developed under the IFERC program of BA activities. The emulsion method can easily produce large volumes of uniform submicron particles. For fusion reactors, tritium breeder pebbles of 1 mm are needed. To examine whether the emulsion method is suitable for pebble fabrication, we used the granulator shown in Fig.9-27.

This granulator comprised two syringes arranged in a T-shaped flow path. One syringe was filled with oil and the other with a  $\text{Li}_{2+x}\text{TiO}_{3+y}$  slurry. The two flow lines from the



**Fig.9-28 Improvement of the grain size of tritium breeder pebbles**

To prevent grain growth,  $\text{Li}_{2+x}\text{TiO}_{3+y}$  gel spheres were sintered in vacuum. The average grain size of the sintered  $\text{Li}_{2+x}\text{TiO}_{3+y}$  pebbles was less than 5  $\mu\text{m}$ .

syringes were connected perpendicular to one another. This arrangement allowed us to cut the  $\text{Li}_{2+x}\text{TiO}_{3+y}$  slurry flow with an oil flow from the oil-filled syringe. The size of the tritium breeder gel spheres was controlled by the flow speeds of the oil and the slurry. The gel spheres were placed in an oil-filled container.

In a series of fabrication trials, the average grain size on the surface and cross section of the sintered  $\text{Li}_{2+x}\text{TiO}_{3+y}$  pebbles was 2–10  $\mu\text{m}$ . Considering the tritium release characteristics, the optimum grain size after sintering should be less than 5  $\mu\text{m}$ . The grain growth factor was assumed to be the presence of binder in the gel spheres. The remaining binder reacted with the Li in  $\text{Li}_{2+x}\text{TiO}_{3+y}$  and generated  $\text{Li}_2\text{CO}_3$ . To prevent this reaction, the  $\text{Li}_{2+x}\text{TiO}_{3+y}$  pebbles were sintered in a vacuum atmosphere (Fig.9-28). The average grain size on the surfaces and cross sections of the sintered  $\text{Li}_{2+x}\text{TiO}_{3+y}$  pebbles was thus less than 5  $\mu\text{m}$ .

The results suggest that the  $\text{Li}_2\text{CO}_3$  generated by the remaining binder affects grain growth and that sintering under vacuum improves the grain size of the pebbles.

### Reference

Hoshino, T. et al., Development of Fabrication Technologies for Advanced Breeding Functional Materials for DEMO Reactors, Fusion Engineering and Design, vol.87, issues 5-6, 2012, p.486-492.