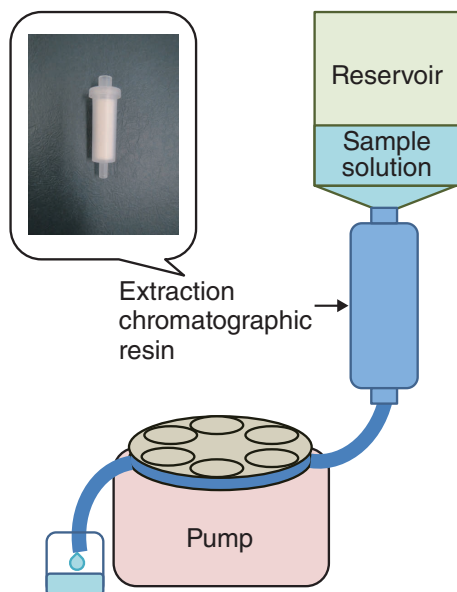


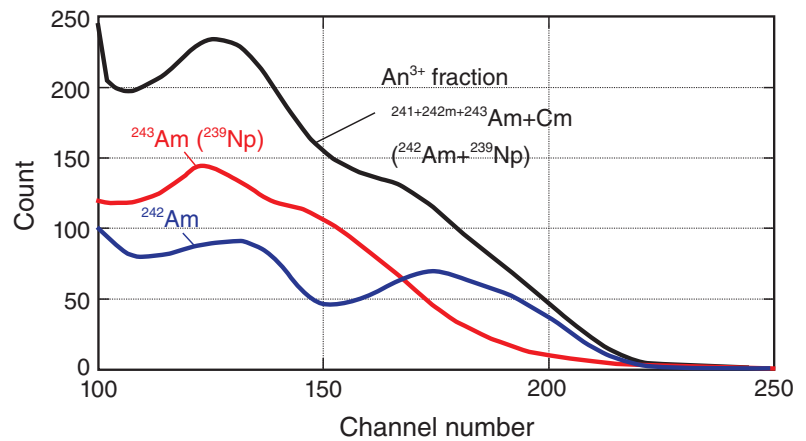
## 8-3 Developing Rapid Method to Analyze Radioactive Waste

### — Developing Method to Analyze $^{242m}\text{Am}$ in Low-Level Radioactive Waste —



**Fig.8-9 System of extraction chromatography**

A cartridge-type of extraction chromatographic resin was used. Solution in a reservoir is pumped from downstream into the cartridge.



**Fig.8-10  $\beta$ -ray spectrum of  $^{242}\text{Am}$  and so on**

$\text{An}^{3+}$  fraction contains  $^{241+242m+243}\text{Am}$  and  $\text{Cm}$ , and  $\beta$ -ray spectrum of  $^{242}\text{Am}$  and  $^{239}\text{Np}$ , which are progeny nuclides of  $^{242m}\text{Am}$  and  $^{243}\text{Am}$ , respectively is obtained.  $^{243}\text{Am}$  source was prepared to estimate  $\beta$ -ray spectrum derived from  $^{239}\text{Np}$ , which was subtracted from the spectrum of the  $\text{An}^{3+}$  fraction to obtain the  $\beta$ -ray spectrum of  $^{242}\text{Am}$ .

The radioactive inventory of low-level radioactive waste (LLW) must be evaluated to dispose of LLW. However, some nuclides that are important for safety are difficult to analyze, such as americium-242m ( $^{242m}\text{Am}$ ). Thermal ionization mass spectrometry (TIMS) is a conventional method to determine the content of  $^{242m}\text{Am}$ .  $^{242m}\text{Am}$  can also be analyzed by  $\alpha$ -ray of curium-242 ( $^{242}\text{Cm}$ ), which is the progeny nuclide of  $^{242m}\text{Am}$  ( $^{242}\text{Cm}$  method). Although the TIMS measuring time is short, more sample is required than for  $^{242}\text{Cm}$  method. However, less sample is available for the  $^{242}\text{Cm}$  method, it takes several months to determine the  $^{242m}\text{Am}$  content. In addition, both methods require separation of Am from Cm.

In the present study, we developed a new method to determine the  $^{242m}\text{Am}$  content that involves measuring  $\beta$ -ray emission from  $^{242}\text{Am}$ , which is the progeny nuclide of  $^{242m}\text{Am}$ . To measure  $\beta$ -rays from  $^{242}\text{Am}$ , Am must be separated from  $\beta$ -ray emitting nuclides in LLW. Therefore, we used a separation method based on extraction chromatography (Fig.8-9). First, trivalent actinides ( $\text{An}^{3+}$ ) and lanthanides ( $\text{Ln}^{3+}$ ) were separated from the major elements in LLW by

using transuranic resin (Eichrom Technologies), which has selectivity for transuranic elements. Next,  $\text{An}^{3+}$  was separated from  $\text{Ln}^{3+}$ , which has similar chemical properties and contains some  $\beta$ -ray emitting nuclides, by using a tetravalent actinide resin (Eichrom Technologies). This  $\text{An}^{3+}$  fraction contains Cm, but the influence of Cm on the  $\beta$ -ray measurement can be subtracted because there is no  $\beta$ -ray emitting Cm in LLW. Conversely,  $\beta$ -rays from neptunium-239 ( $^{239}\text{Np}$ ), which is a progeny nuclide of  $^{243}\text{Am}$ , have to be detected and considered. Therefore, a  $^{243}\text{Am}$  source was prepared to measure  $\beta$ -ray spectra of  $^{239}\text{Np}$ , which were subtracted from those of  $\text{An}^{3+}$  to obtain the radioactivity of  $^{242}\text{Am}$  (Fig.8-10). As a verification, the value obtained was compared with that determined by the  $^{242}\text{Cm}$  method.

The separation step in the new method to determine the radioactivity of  $^{242}\text{Am}$  is simpler than that for the TIMS and  $^{242}\text{Cm}$  methods. Less sample is required by this new method than by the TIMS method and is comparable to that required by the  $^{242}\text{Cm}$  method. The measuring time for this new method is considerably shorter than that for the  $^{242}\text{Cm}$  method.

#### Reference

Shimada, A. et al., A New Method to Analyze  $^{242m}\text{Am}$  in Low-Level Radioactive Waste Based on Extraction Chromatography and  $\beta$ -ray Spectrometry, Analytical Chemistry, vol.85, no.16, 2013, p.7726-7731.