

1-11 Transfer of Radioactive Cesium in the Hardwood Forest

— Study of Radioactive-Cesium-Transport Behavior in Trees in a Forest —

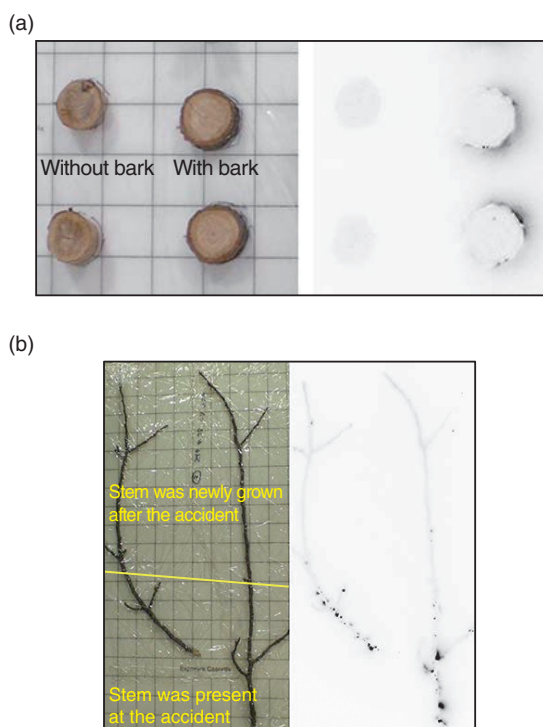


Fig.1-28 Autoradiograph in a deciduous broad-leaved tree

(a) Radioactive Cs was found mostly on the bark. (b) Radioactive Cs was homogeneously distributed in the stem that was newly grown since the 1F accident, whereas a heterogeneous and spotted radioactive-Cs distribution was observed in the stem that was present at the time of the accident.

Numerous radionuclides were released into the environment as a result of the accident at the TEPCO's Fukushima Daiichi NPS (1F), which occurred as a result of the Great East Japan Earthquake on March 11, 2011. Radioactive cesium (Cs) isotopes with relatively long half-lives are still present in the environment. Deciduous broad-leaved trees such as Konara oak, Mizunara oak, and Japanese chestnut are used in people's daily lives as forest products (firewood and mushroom cultivation). From the perspective of the use of forest products, it is important to understand the radioactive Cs behavior in these trees. At the time of the 1F accident, because the deciduous trees were in a dormant stage and their leaves were falling, released radioactive Cs was caught in the branches and trunks. In cases where the radioactive Cs deposited onto the surface of a tree was transferred inside, some of the radioactive Cs was transferred to newly generated branches and leaves after the 1F accident. In contrast, the radioactive Cs on the trees' outside can be transferred by rain running down the trunk (stem flow). To elucidate the distribution of radioactive Cs and the radioactive-Cs-transfer mechanism to stemflow in deciduous broad-leaved trees, we surveyed the radioactive-Cs distribution and concentration in the stems of Chestnut trees growing in deciduous broad-leaved forest located approximately 35 km from 1F as well as the radioactive-Cs concentration and the

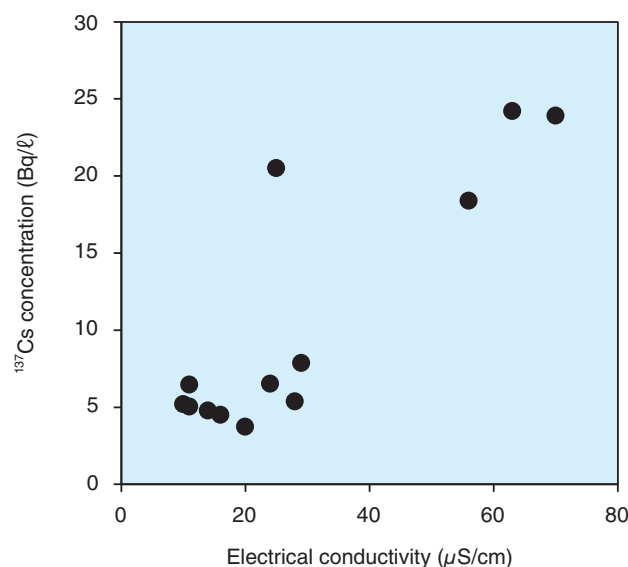


Fig.1-29 Relationship between radioactive Cs (¹³⁷Cs) concentration and electrical conductivity in a dissolved fraction of stemflow

A positive correlation was observed between electrical conductivity and ¹³⁷Cs concentration in the dissolved fraction of the stemflow. This result suggests that major ions and radioactive Cs in the stemflow are controlled by the same elution mechanism.

characteristics of the soluble fraction (the solution that passed through a filter with a pore size of 0.45 μm) of stemflow.

On the bark surface of the chestnut trees that was presented at the time of the 1F accident, the radioactive Cs had a heterogeneous and spotty distribution. In contrast, radioactive Cs on branches newly grown after the 1F accident was uniformly distributed (Fig.1-28). The radioactive-Cs concentration in the bark was approximately 10 times that of rest of the wood. The radioactive-Cs concentration in the dissolved fraction of the stem flow was approximately 10⁻³ times that on the bark. Because radioactive-Cs concentration in the bark is still high and that in the soluble fraction of stem flow is low, elution of radioactive Cs from the trees to the stem flow would occur very slowly. Electrical conductivity is an index of the flowability of electricity in the solution and depends upon the abundances of ions in solution. There was a positive correlation between the radioactive-Cs concentration and the electrical conductivity of the dissolved fraction of the stem flow. This result suggests that the predominant ions (nitrogen, phosphorus, sulfur, potassium, calcium, magnesium, manganese, etc.) and radioactive Cs in the stemflow are controlled by the same elution mechanism (Fig.1-29).

We need to study the elution mechanism of radioactive Cs and other ions into the stem flow.

Reference

Sasaki, Y. et al., The Transfer of Radiocesium from the Bark to the Stemflow of Chestnut Trees (*Castanea Crenata*) Contaminated by Radionuclides from the Fukushima Dai-Ichi Nuclear Power Plant Accident, Journal of Environmental Radioactivity, vol.161, 2016, p.58-65.