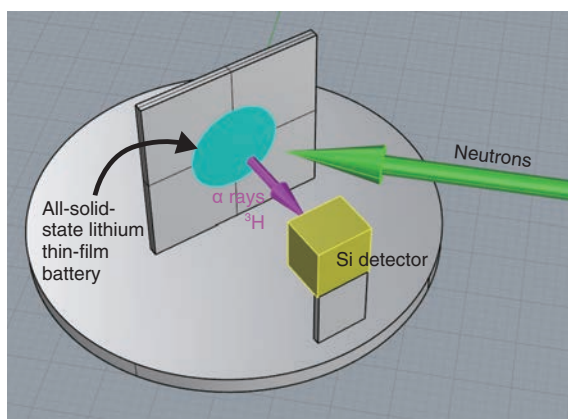


## 5-5 Visualizing Lithium-Ion Transfer in All-Solid-State Batteries — Neutron Depth Profiling for the Operand Measurement of Li Ions —

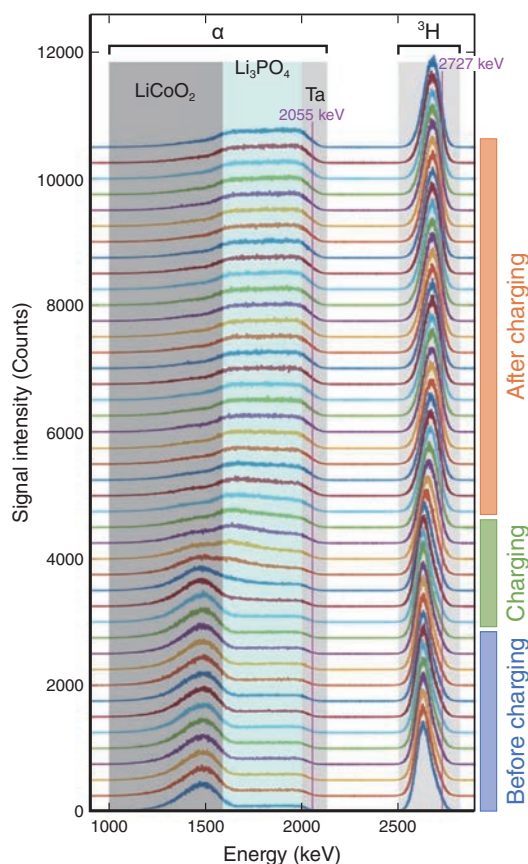


**Fig.1 Overview of neutron depth profiling**

When an all-solid-state Li thin-film battery is irradiated with neutrons, a nuclear reaction occurs between the  ${}^6\text{Li}$  present in the battery and the neutrons, and  $\alpha$ -rays and tritium ( ${}^3\text{H}$ ) are emitted. The energy of the generated  $\alpha$ -rays and tritium is measured with a Si detector. This experiment was conducted in the research reactor JRR-3.

**Fig.2 Energy spectra of neutron depth profiling**

The spectrum changes before, during, and after charging are shown. Higher energies indicate that the Li ions are closer to the surface. Li ions are transferred from the cathode ( $\text{LiCoO}_2$ ) to the electrolyte ( $\text{Li}_3\text{PO}_4$ ) because of charging. The original energies of the  $\alpha$ -ray and  ${}^3\text{H}$  are 2055 keV and 2727 keV, respectively. Scan the QR code to watch the movie.



Lithium-ion batteries (LIBs) can store large amounts of electricity and are used in a variety of devices such as smartphones. However, they use organic solvents as the electrolyte, which poses the risk of ignition. In contrast, all-solid-state LIBs are batteries with a solid electrolyte. They are expected to become the next generation of batteries. Compared to the conventional lithium-ion batteries, all-solid-state lithium-ion batteries offer the benefits of higher energy density, safety, and durability, but this technology is still in its infancy and has not yet reached mass production.

To improve the performance of all-solid-state LIBs, it is important to clarify the migration mechanism of lithium ions inside the LIB. However, visualizing this migration process is a challenge. In particular, it has been difficult to track the movement of lithium ions in real time during charging and discharging.

Therefore, we challenged ourselves to observe the lithium ions in all-solid-state LIBs by neutron depth profiling. Lithium has two isotopes with different weights ( ${}^6\text{Li}$  and  ${}^7\text{Li}$ );  ${}^6\text{Li}$  reacts very well with neutrons. When neutrons react with  ${}^6\text{Li}$  in the battery,  $\alpha$  rays and tritium ( ${}^3\text{H}$ ) are emitted through a thermal neutron-induced nuclear reaction (Fig.1). By measuring the energy spectrum of the generated  $\alpha$ -particles and tritium, we can visualize the depth from the surface of the reacting lithium

from the energy loss, since the energy is lost depending on the depth at which they were generated.

Experiments were conducted using the prompt  $\gamma$ -ray analyzer (PGA) installed in the research reactor JRR-3. In this study, all-solid-state LIB samples were prepared using a  ${}^6\text{Li}$ -enriched cathode (lithium cobaltate) to increase the analytical sensitivity. In nature,  ${}^6\text{Li}$  comprises only 7.5%, but in our samples, it was enriched to 95.4%. Lithium phosphate was used as the solid electrolyte, and tantalum as the negative electrode. The battery sample was placed in a vacuum container and irradiated with thermal neutrons for neutron depth profiling. Through this experiment, we succeeded, for the first time in the world, in dynamically capturing the lithium-ion migration process in all-solid-state LIBs while actually charging the battery (Fig.2). A comparison of the experimental results with simulations revealed that the lithium ions in the solid electrolyte migrate to fill vacancies and that they do not migrate throughout the entire area of the solid electrolyte but only in specific areas within the solid electrolyte.

The development of technology to visualize the behavior of lithium ions in all-solid-state LIBs during charging and discharging is expected to advance the development of such batteries in the future.

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### Reference

Kobayashi, T., Osawa, T. et al., In-Operando Lithium-Ion Transport Tracking in an All-Solid-State Battery, *Small*, vol.18, issue 46, 2022, 2204455, 9p.